

# Local heat transfer of compressible fluid in porous media: application to the HBC fuse

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## Abstract

We propose a new model to describe compressible fluid flows in porous media introducing a microscopic thermal model to evaluate the local thermal evolution of the solid phase. Gas flow in porous media is governed by the homogenized Euler equations. We first introduce the classical non-equilibrium thermal model where the porous medium thermal distribution is driven by the heat equation. We then present the new thermal modelling introducing a characterization of the depth of the porous medium deepness. To compare the models, we have performed two critical tests: the heat exchanger and the explosion. We show that both the models agree in the first test but differ in the second case.

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**Keywords:** Porous media; Heat transfer; Finite volume; Thermal local non-equilibrium

## 1. Introduction

The high breaking capacity (HBC) fuse involves heat transfer in porous media with high pressure and velocity for the gas flow (Jakubiuk and Lipski, 1993). During the discharge process (about 5 ms), a 20000 K plasma at  $10^6$  Pa is generated in the fuse core and spreads in a silica sand with gas velocities up to  $300 \text{ ms}^{-1}$  (Bussière, 2001). Heat transfer between the hot gas and the porous medium warms and partially vaporizes the silica grains. Many investigations of gas flow in porous media concern heat exchange where the gas velocities are low, the temperature gradient is weak and the process is slow compared to the fuse process characteristics.

In the heat exchange, the pressure and Darcy force mainly govern the flow, the diffusion and the heat trans-

fer mainly determine the temperature distribution. From the energy equation point of view, classical methods consider two different models in theoretical and numerical research: the local thermal equilibrium model (Jiang et al., 1999a) and the local thermal non-equilibrium model (Hsieh and Lu, 2000; Amiri and Vafai, 1994; Lee and Vafai, 1999).

The local thermal equilibrium model assumes that the solid phase temperature is equal to the gas temperature leading to an important simplification of the mathematical model. Nevertheless, the assumption of local thermal equilibrium between the gas and the solid phases is inadequate for a large number of problems. The local thermal non-equilibrium model has received considerable attention where we assume a temperature difference between the solid and the gas phases. Two equations govern the solid and the gas temperature coupled by a heat transfer term.

Both the models assume that the temperature distribution inside the particles of the porous medium is

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## Nomenclature

$A_0$	specific surface area ( $\text{m}^{-1}$ )
$c_p, c_v$	specific heat ( $\text{J kg}^{-1} \text{K}^{-1}$ )
$d_p$	particle diameter (m)
$e$	specific internal energy ( $\text{J kg}^{-1}$ )
$E$	total energy ( $\text{J m}^{-3}$ )
$h_{sf}$	heat transfer coefficient ( $\text{W m}^{-2} \text{K}^{-1}$ )
$K$	permeability ( $\text{m}^2$ )
$k$	thermal conductivity ( $\text{W m}^{-1} \text{K}^{-1}$ )
$Nu$	Nusselt number
$p$	pressure (Pa)
$Pr$	Prandtl number
$Re$	Reynolds number
$T$	temperature (K)
$u$	interstitial velocity ( $\text{m s}^{-1}$ )
$Q$	heat transfer ( $\text{W m}^{-3}$ )
$U$	vector of independent variables

## Greeks

$\beta$	Forchheimer coefficient ( $\text{m}^{-1}$ )
$\gamma$	adiabatic index
$\phi$	porosity
$\mu$	gas viscosity ( $\text{kg m}^{-1} \text{s}^{-1}$ )
$\rho$	density ( $\text{kg m}^{-3}$ )

## Subscripts

$g$	gas
$geff$	effective property for gas
$in$	channel inlet
$out$	channel outlet
$s$	solid
$seff$	effective property for solid
$sf$	interfacial conditions

homogeneous, i.e. the peripheral temperature and core temperature of the grain are equal; but this assumption is not realistic in the case of the HBC fuse. As the classical models have failed, our specific application requires a customised model to take into account the non-homogeneous distribution of the temperature in the grain. For example, the solid grain has an important temperature gradient and its peripheral layer can reach the fusion temperature since the core remains cold. The main contribution of this present paper is to provide a modelling of the local energy distribution in the porous medium to take account of the thermal transition between the core and the surface of the grains.

A similar local heat exchange model has already been considered (Rohsenow et al., 1998) for dispersed particles in flow where we compute the thermal distribution in each sphere. Such a model considers that each particle is isolated one from the other. In the fuse case, the particles are packed and their geometry is not spherical. We thus propose a new local heat transfer model to characterize locally the temperature distribution of the porous medium by introducing a characteristic variable  $\xi$  which parameterizes the local depth in the porous medium.

The structure of the paper is as follows: in Section 2, we introduce the classical thermal non-equilibrium model comprising the Euler equations for the gas phase and the heat equation for the solid phase. We then present in Section 3 a new model to evaluate more accurately the local solid temperature distribution using a microscopic description at the pore scale and to determine the heat transfer between the two phases.

Finally, we test two characteristic situations: the heat exchange and the explosion case where we illustrate the difference between the two models.

## 2. The classical two-phase model

### 2.1. Mathematical modelling

We consider a compressible gas flowing through a one-dimensional porous medium composed of silica sand of mean radius  $R$  (see Fig. 1). Fluid flow enters the medium with a constant velocity  $u_0$ , pressure  $P_0$  and temperature  $T_0$  and the outflow is at atmospheric pressure.

We assume that all the functions only depend on the time  $t$  and the one-dimensional variable  $x$ . The governing equations for a single-phase fluid flow in an isotropic, homogeneous porous medium are derived from the Euler equations (Jiang and Ren, 2001; Jiang et al., 1999a).

Bear and Bachmat (1990) give the basic equations for an incompressible fluid in a porous medium and Nield (1994) gives the extension to the compressible case. The homogenization process leads us to consider three major terms: the viscous interaction between the gas and the porous medium, the heat transfer, and the momentum and thermal dispersions.

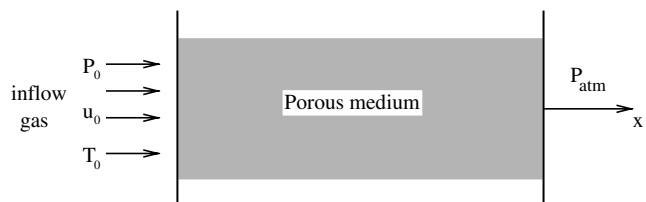


Fig. 1. Schematic diagram of a gas flowing through a porous medium.

The homogenization of the viscous term provides the Darcy force. Bear and Bachmat obtain the expression both for the isochoric and non-isochoric case (Eqs. (2.6.43) and (2.6.42) in their book). It turns out that the Darcy term is the same in both cases where an additional macroscopic viscous term is only added in the compressible case. On the other hand, in the case of a high speed flow, the laminar flow goes into a Forchheimer flow regime and a new force is added to the Darcy force to describe the fluid–structure interaction. An explicit Forchheimer force expression for incompressible fluid has been obtained by homogenization (Skjetne and Auriault, 1999; Ochoa-Tapia and Whitaker, 1998; Whitaker, 1972). In Chen et al. (2001), the authors predict that the Forchheimer law is valid even for compressible flow since it remains incompressible at a local level (De Ville, 1996).

To take into account of the thermal dispersion, Sozen and Vafai introduce in a series of papers (Alazmi and Vafai, 2000; Amiri and Vafai, 1994; Amiri et al., 1995; Sözen and Vafai, 1990; Vafai and Sözen, 1990a; Vafai and Sözen, 1990b) a diffusion term characterized by an efficient coefficient linked to the Reynolds and Prandtl numbers. The same technique is used for the momentum dispersion where Bear and Bachmat (1990) propose an artificial viscosity for isochoric flow characterized by a mechanical dispersion coefficient linked to the Peclet number. For the turbulent gas with dispersed particles, Simonin (1991) uses the artificial viscosity concept taking account of the turbulent fluid motion on a large scale with a  $q^2-\epsilon$  model (Balzer et al., 1995).

Finally, a source term depending on the temperature difference between the two media is added to the fluid and porous medium energy equations to describe the heat transfer between the fluid and the porous medium. Several correlations are proposed to determine the heat transfer coefficient depending on the Reynolds, Nusselt and Prandtl numbers.

In the present paper, we neglect the macroscopic viscous term and the momentum dispersion term since the fluid is essentially governed by the convection and the pressure. On the other hand, we characterize the thermal dispersion using a diffusion operator (Amiri and Vafai, 1994) in order to measure its impact on the thermal model. Consequently, the considered model takes the form:

$$\frac{\partial(\rho_g \phi)}{\partial t} + \frac{\partial(\rho_g \phi u)}{\partial x} = 0, \quad (1)$$

$$\frac{\partial(\rho_g \phi u)}{\partial t} + \frac{\partial(\rho_g \phi u^2 + \phi P)}{\partial x} = -\phi^2 \frac{\mu}{K} u - \phi^3 \beta \rho_g |u| u, \quad (2)$$

$$\frac{\partial(\phi E)}{\partial t} + \frac{\partial[(E + P) \cdot \phi u]}{\partial x} = \frac{\partial}{\partial x} \left( k_{\text{geff}} \frac{\partial T_g}{\partial x} \right) - Q, \quad (3)$$

where  $\rho_g$  is the gas density,  $u$  the interstitial velocity,  $P$  the pressure and  $\phi$  the porosity. Note that we keep the advective and the inertial terms in the momentum equation leading to a hyperbolic system. In low velocity flow, such terms are omitted (Alazmi and Vafai, 2000; Amiri and Vafai, 1994). On the right-hand side, the mechanical interaction is represented by the viscous force (Darcy law), with  $K$  the permeability and  $\mu$  the gas viscosity and the inertial resistance (Forchheimer law) where  $\beta$  is the forchheimer coefficient (Macdonald et al., 1979; Teng and Zhao, 2000; Whitaker, 1996). In the energy conservation equation,  $E$  is the gas total energy per unit volume and  $e$  is the specific internal energy:

$$E = \rho_g \left( \frac{1}{2} u^2 + e \right) \quad \text{with} \quad e = c_v T_g,$$

where  $c_v$  is the gas specific heat coefficient at constant volume and  $T_g$  is the gas temperature. In addition, to close the system of equations, we use the ideal gas equation of state

$$P(\rho, e) = (\gamma - 1) \rho_g e \quad \text{with} \quad \gamma > 1, \quad (4)$$

where  $\gamma$  is the ratio of specific heat coefficients of the gas. Ishii (1975) presents a general method to get the state equation. Eq. (4) is not correct with the averaged values since additional terms like the turbulent kinetic energy appear with the averaged process. But, if we neglect the fluctuating moments in comparison with the averaged values, we conclude that state equation relation (4) with the averaged variables is a rather good approximation.

The thermal diffusion is driven by the gas phase effective thermal coefficient  $k_{\text{geff}}$  given by the empirical correlation of Amiri and Vafai (1994):

$$k_{\text{geff}} = \phi k_g + 0.5 \left[ Pr \left( \frac{\rho u d_p}{\mu} \right) \right] k_g. \quad (5)$$

The thermal interaction  $Q = h_{\text{sf}} A_0 (T_g - T_s)$  between the gas and the solid grains is governed by an empirical law (Rohsenow et al., 1998) which depends on the convective heat transfer coefficient  $h_{\text{sf}}$ , the specific interface surface area per unit volume  $A_0 = 6(1 - \phi)/d_p$  (Jiang and Ren, 2001) where  $d_p$  is the particle diameter, and the gas and solid temperatures. The convective heat transfer coefficient  $h_{\text{sf}}$  has been determined experimentally by various authors. Wakao and Kaguei (1982) (see also Achenbach, 1995) propose a correlation in the case of packed spheres:

$$h_{\text{sf}} = \frac{k_g}{d_p} (2 + 1.1 Re^{0.6} Pr^{1/3}). \quad (6)$$

Other correlations are used, for example Alazmi and Vafai (2000) propose

$$h_{sf} = \left[ \frac{d_p \phi}{0.2555 Pr^{1/3} Re^{2/3} k_g} + \frac{d_p}{10 k_g} \right]^{-1}, \quad (7)$$

where  $Nu = h_{sf} d_p / k_g$ ,  $Re = \rho_g \phi u d_p / \mu$  and  $Pr = \mu c_{pg} / k_g$  are respectively the Nusselt, Reynolds and Prandtl numbers,  $k_g$  is the gas thermal conductivity. We can notice that for large Reynolds numbers, all the expressions give the same heat transfer coefficient. The correlations only differ for low Reynolds numbers.

Finally, to model the heat transfer process between the gas and solid phases, we introduce the classical heat equation with a longitudinal thermal conduction in the solid phase (Bortolozzi and Deiber, 2001) to evaluate the solid temperature distribution  $T_s$ :

$$\rho_s c_{vs} \frac{\partial T_s}{\partial t} = \frac{k_{seff}}{(1-\phi)} \frac{\partial^2 T_s}{\partial x^2} + \frac{Q}{(1-\phi)}, \quad (8)$$

where  $\rho_s$  denotes the solid density,  $c_{vs}$  is the solid specific heat at constant volume and  $k_{seff}$  is the porous medium conductivity with  $k_{seff} = (1-\phi)k_s$ .

## 2.2. Discretization schemes and numerical method

At time  $t = 0$ , we assume that the gas is at rest, the pressure is atmospheric and the temperatures (solid and gas) are equal to 300 K. For  $t > 0$ , we inject a hot gas with a temperature  $T_0$ , a pressure  $P_0$  and a velocity  $u_0$  at  $x = 0$ . On the other hand, we prescribe at  $x = L$  an atmospheric pressure outflow.

Most authors use the method of finite differences since they do not take the advective term into account. In the present case, we use a finite volume scheme to solve numerically the gas system (Eqs. (1)–(3)). The motivation is to provide a correct resolution of the hyperbolic system without adding any artificial viscosity to stabilize the scheme. We write the system in the conservative form (Toro, 1997):

$$\frac{\partial U}{\partial t} + \frac{\partial F(U)}{\partial x} = S(U), \quad (9)$$

where  $U$  is the conservative variable vector and  $F(U)$ ,  $S(U)$  are respectively the flux vector and the source vector in function of vector  $U$  and the fixed parameters:

$$U = \begin{pmatrix} \rho_g \phi \\ \rho_g \phi u \\ \phi E \end{pmatrix}, \quad F(U) = \begin{pmatrix} \rho_g \phi u \\ \rho_g \phi u^2 + \phi P \\ (E + P) \phi u \end{pmatrix},$$

$$S(U) = \begin{pmatrix} 0 \\ -\phi^2 \frac{\mu}{k} u - \phi^3 \beta \rho_g |u| u \\ \frac{\partial}{\partial x} \left( k_{seff} \frac{\partial T_g}{\partial x} \right) - h_{sf} A_0 (T_g - T_s) \end{pmatrix}.$$

### 2.2.1. The fractional step technique

In order to obtain an approximate solution of system (9), we use a fractional step technique (Ben-Dor et al.,

1997; Rochette and Clain, 2003; Rochette and Clain, 2004): for a given time interval  $[t^n, t^{n+1}]$ , we first solve the homogeneous conservative system and then the right-hand side terms. More precisely, let  $U^n$  be an approximation of  $U(t^n)$  at time  $t^n$ , in order to obtain an approximation of  $U(t^{n+1})$  at time  $t^{n+1} = t^n + \Delta t$ , we first determine an approximate solution over the time interval  $[t^n, t^{n+1}]$  of the homogeneous problem

$$\begin{cases} \frac{\partial U}{\partial t} + \frac{\partial F(U)}{\partial x} = 0, \\ U(x, t^n) = U^n, \end{cases} \quad (10)$$

using the finite volume scheme presented in Section 2.2.2. Let us now assume that  $\tilde{U}^{n+1}$  is the approximate solution value at  $t = t^{n+1}$  of Eq. (10), we solve the ordinary differential equation

$$\frac{dU}{dt} = S(U), \quad (11)$$

using  $\tilde{U}^{n+1}$  as initial condition at  $t = t^n$ . The solution at  $t = t^{n+1}$  of Eq. (11) is an approximation  $U^{n+1}$  of  $U(t^{n+1})$ .

### 2.2.2. The homogeneous and source problems

We briefly describe the finite volume method based on a Roe's solver (Roe, 1981). Let us consider a uniform grid of the porous domain  $[0, L]$  setting  $x_i = (i - \frac{1}{2})\Delta x$  with  $i = 1 \dots N$ , and

$$x_{i-1/2} = x_i - \frac{\Delta x}{2}, \quad x_{i+1/2} = x_i + \frac{\Delta x}{2}.$$

For each index  $i$ ,  $C_i = [x_{i-1/2}, x_{i+1/2}]$  represents the cell of center  $x_i$  and length  $\Delta x$ . Assuming that we know an approximation  $U_i^n = ((\rho_g \phi)_i^n, (\rho_g \phi u)_i^n, (E \phi)_i^n)$  of  $U(x_i, t^n)$  at the time  $t^n$  on each cell  $C_i$ , we seek an approximation  $\tilde{U}^{n+1}$  for the homogeneous problem (10) using an explicit finite volume scheme of the form

$$\tilde{U}_i^{n+1} = U_i^n - \frac{\Delta t_h}{\Delta x} (F_{i+1/2}^n - F_{i-1/2}^n), \quad (12)$$

where  $F_{i+1/2}^n$  and  $F_{i-1/2}^n$  represent respectively the numerical fluxes calculated at the interface cells  $x = x_{i+1/2}$  and  $x = x_{i-1/2}$  using the Roe's solver. We reach second order spatial accuracy using the MUSCL method with slope limited (Yee, 1987; Abgrall, 1988). Moreover, the time step  $\Delta t_h$  is chosen in agreement with the Courant–Friedrichs–Lewy (CFL) condition (Courant and Friedrichs, 1948).

The second step consists in solving the first-order system of ordinary differential equations (Toro, 1997):

$$\frac{dU}{dt} = \begin{pmatrix} 0 \\ -\phi^2 \frac{\mu}{k} u - \phi^3 \beta \rho_g |u| u \\ -h_{sf} A_0 (T_g - T_s) \end{pmatrix}. \quad (13)$$

Numerically, we use a fourth-order explicit Runge–Kutta method. Since the method is explicit, a stability condition arises and the time step  $\Delta t$  must be lower than a

limiting time interval  $\Delta t_i$ . Finally, we add the diffusion contribution by:

$$\rho_g c_{pg} T_{g,i}^{n+1} = \rho_g c_{pg} T_{g,i}^n + \frac{\Delta t_d}{\Delta x} (f_{i+1/2}^n - f_{i-1/2}^n), \quad (14)$$

where  $\Delta t_d$  is the time step of the diffusion term such that  $\Delta t_d < \rho_g c_{pg} (\Delta x)^2 / k_{\text{seff}}$  to satisfy the stability condition of the explicit scheme. The computation of  $f_{i+1/2}^n$ ,  $f_{i-1/2}^n$  is detailed in the next section.

### 2.2.3. Solid energy equation

To compute the temperature in the solid, we consider the same uniform grid of the domain  $[0, L]$ . Assuming that we know an approximation  $T_{s,i}^n$  of  $T_s(x_i, t^n)$  at  $t^n$  on the cell  $C_i$ , we seek an approximate  $T_{s,i}^{n+1}$  at time  $t^{n+1}$  using an explicit finite volume method given by:

$$\begin{aligned} \rho_s c_{vs} T_{s,i}^{n+1} &= \rho_s c_{vs} T_{s,i}^n + \frac{\Delta t_s}{\Delta x} (f_{i+1/2}^n - f_{i-1/2}^n) \\ &+ \Delta t_s h_{s,i}^n A_0 (T_{g,i}^n - T_{s,i}^n), \end{aligned} \quad (15)$$

where central differences are used to determine the fluxes, i.e.:

$$f_{i+1/2}^n = k_{\text{seff}} \frac{T_{s,i+1}^n - T_{s,i}^n}{\Delta x} \quad \text{and} \quad f_{i-1/2}^n = k_{\text{seff}} \frac{T_{s,i}^n - T_{s,i-1}^n}{\Delta x}.$$

The explicit scheme is stable if we satisfy the condition  $\Delta t_s < \rho_s c_{vs} (\Delta x)^2 / k_{\text{seff}}$  but in practice, the CFL condition for the gas flow problem gives a more restrictive condition than the stability condition. Therefore an implicit scheme is not necessary.

### 2.2.4. The numerical treatment of the boundary conditions

Boundary conditions are necessary to complete the flux contribution on the interfaces  $x = 0$  and  $x = L$ . We add fictitious cells  $C_0$  and  $C_{N+1}$  where we prescribe the boundary conditions (Toro, 1997). For the  $C_0$  cell, we set the inflow conditions

Table 1

Input data used in the numerical calculation

Porosity	$\phi = 0.4$
Permeability	$K = 6 \times 10^{-11} \text{ m}^2$
Forchheimer coefficient	$\beta = 3.3 \times 10^4 \text{ m}^{-1}$
Grain diameter	$d_p = 3 \times 10^{-4} \text{ m}$
Viscosity	$\mu = 1.8 \times 10^{-5} \text{ kg m}^{-1} \text{ s}^{-1}$
Gas thermal conductivity	$k_g = 0.026 \text{ W m}^{-1} \text{ K}^{-1}$
Porous media thermal conductivity	$k_{\text{seff}} = 0.33 \text{ W m}^{-1} \text{ K}^{-1}$
Ratio of specific heat	$\gamma = 1.4$
Solid specific heat at constant volume	$c_{vs} = 1420.74 \text{ J kg}^{-1} \text{ K}^{-1}$

$$U_0^n = \begin{pmatrix} (\rho_g \phi)_0^n \\ (\rho_g \phi u)_0^n \\ (E\phi)_0^n \end{pmatrix},$$

and for the  $C_{N+1}$  cell, we set the outflow conditions under the assumption that the outside pressure is atmospheric

$$U_{N+1}^n = \begin{pmatrix} (\rho_g \phi)_{N+1}^n \\ (\rho_g \phi u)_{N+1}^n \\ (E\phi)_{N+1}^n \end{pmatrix} = \begin{pmatrix} (\rho_g \phi)_N^n \\ (\rho_g \phi u)_N^n \\ \frac{P_{\text{atm}}}{(\gamma-1)} + \frac{1}{2} (\rho_g \phi u)_N^n u_N^n \end{pmatrix}.$$

### 2.3. Convergence tests

We consider a 0.1 m size domain filled with a porous medium composed of silica sand. We inject a constant gas mass flux  $(\rho_g \phi u)_0^n = 25 \text{ kg m}^{-2} \text{ s}^{-1}$  with a temperature of  $(T_g)_0 = 1160 \text{ K}$  during  $t = 1 \text{ s}$ . At the initial time  $t = 0$ , the system is at rest,  $u = 0 \text{ m s}^{-1}$ , the gas present in the porous medium interstices is at atmospheric pressure and ambient temperature  $T_g = T_s = 300 \text{ K}$ . The gas and the porous medium parameters used in the numerical simulations are shown in Table 1.

In order to compare the two models and for the sake of simplicity, we have used constant coefficients. The numerical solver of the gas system has been tested in a similar situation in Rochette and Clain (2004) where

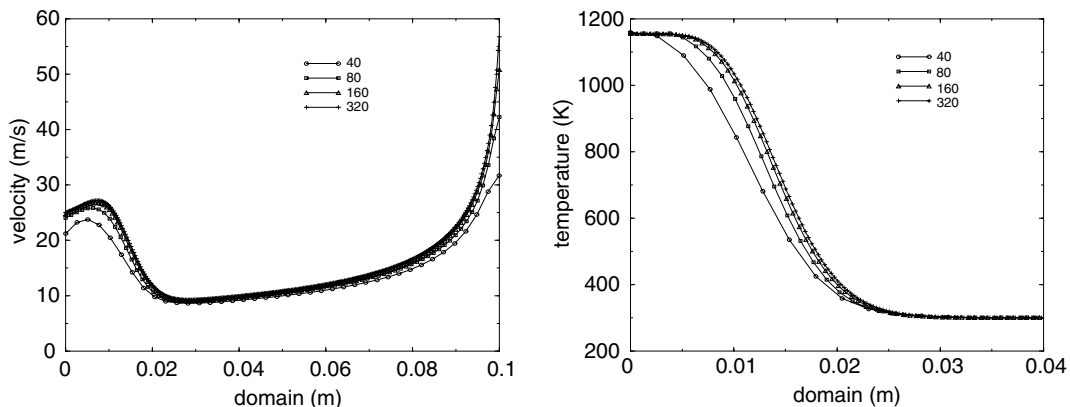


Fig. 2. Velocity and temperature distribution in the porous medium domain at  $t = 1 \text{ s}$  for several meshes.



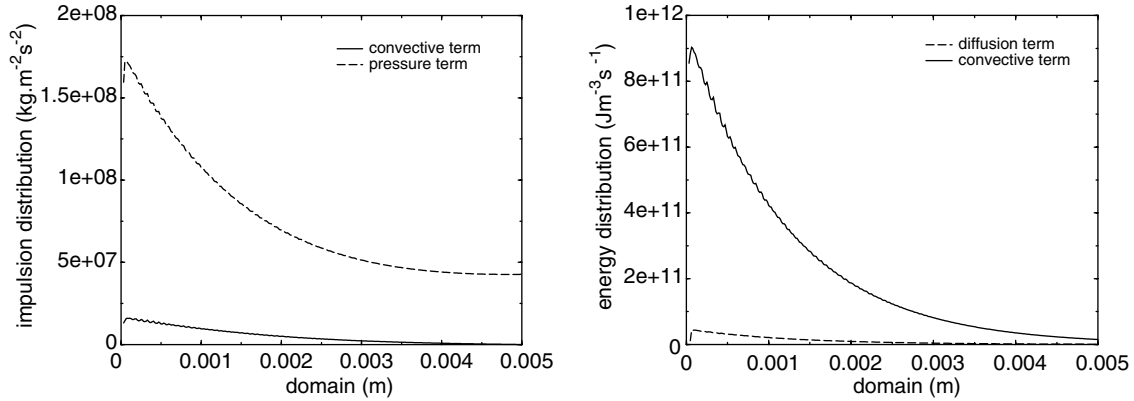


Fig. 3. (left) Comparison of the convective and pressure forces at  $t = 5$  ms. (right) Comparison of the energy convection and energy diffusion at  $t = 5$  ms.

we present a convergence test based on a stationary solution. All the computations have been performed with a C++ code using a dedicated Object Finite Element Library OFELI (Touzani, 1998).

We present simulation results with four meshes composed of 40, 80, 160 and 320 uniform cells respectively. Fig. 2 represents the evolution of the gas velocity and gas temperature in the domain after 1 s. For the four different meshes, the results are similar with a significant difference between 40 and 320 cells close to the boundaries. The stability condition involves four time steps  $\Delta t_h$ ,  $\Delta t_l$ ,  $\Delta t_d$  and  $\Delta t_s$ , so we choose a time step  $\Delta t$  lower than  $\min(\Delta t_h, \Delta t_l, \Delta t_d, \Delta t_s)$ . In our application, the minimum is the homogeneous time step  $\Delta t_h$ .

We note that for a mesh of 80 cells, the convergence is sufficient in comparison with a mesh of 320 cells and in the sequel, we will use an 80 cell mesh to reduce the time computation. Another remark is the high mesh sensibility of the outflow velocity to respect the outside atmospheric pressure condition. Fortunately, it has a low impact from the energy point of view. We shall use this benchmark to compare the numerical solution of the micro/macro model presented in the Section 3.

#### 2.4. Simulations and justifications

A first set of simulations has been performed to compare the correlations used to characterize the heat transfer. For the same initial conditions given in the previous section, we run the program with both the Wakao and Kagui (1982) (Eq. (6)) formula and the correlation proposed by Alazmi and Vafai (2000) (Eq. (7)). We do not present any curve since we obtain the same thermal distributions. Indeed, the relations give the same heat transfer coefficient for high Reynolds numbers (i.e.  $Re \geq 300$ ) and the heat transfer contribution is negligible for low Reynolds numbers. Consequently, we shall use the Wakao and Kagui (1982) formula in the sequel.

Most authors do not consider the convection term in the momentum equation since the pressure forces is

greater than the convection force. This is realistic for the heat exchanger case. We compare here the convective term  $\partial(\rho u^2)/\partial x$  and the pressure  $\partial p/\partial x$  forces and compute their contributions. In a first run, we use the initial and boundary conditions of the previous section. We obtain velocity flow higher than  $20 \text{ ms}^{-1}$ . In this case, the convective term represents less than 1% of the pressure force and can be neglected. We perform a second run using boundary conditions such that we obtain a gas velocity of about  $200 \text{ ms}^{-1}$  which is realistic for the HBC fuses. In this case, the convective contribution is greater than 10% of the pressure force and should not be omitted (see Fig. 3). We proceed in the same way to evaluate the thermal diffusion contribution and compare the numerical values of the convective part  $\partial(u(E + p))/\partial x$  with the diffusion part  $\partial(k_{\text{eff}} \partial T_g / \partial x) / \partial x$ . In the first run (low velocity), the result is that the diffusion is smaller (<1%) than the convective part. We have also run the code without the diffusion and the temperature, velocity and pressure are the same. In the second run (high velocity), we get that the diffusion contribution is about 5% (see Fig. 3).

### 3. The Macroscopic model with a microscopic 2D approach

As we have emphasized in the introduction, the temperature inside the grain is not homogeneous, which leads us to consider a new model in which we predict the temperature distribution in the grains. A first idea consists of representing the grain by a sphere of radius  $R$  and solving the heat equation adding some flux condition on the boundary as Rohsenow et al. (1998), Section 9.34. This model is correct if the spheres (i.e. the grains) are isolated from one another, but in the HBC fuse (or any packed porous medium), the grains are in contact and their geometry is not spherical, therefore the sphere model is inapplicable. To model the heat distribution in the grains, we introduce a characteristic variable  $\xi$  such

that  $\xi = 0$  corresponds to the center of the grain and  $\xi = l_1$  to the boundary where we shall set the  $l_1$  value in the sequel. At any point  $x$ , we give a local description of the solid temperature with a temperature  $\theta_s(x, \xi)$  governed by the heat equation:

$$\frac{\partial \epsilon_s}{\partial t} - \frac{\partial}{\partial \xi} \left( k_s \frac{\partial \theta_s}{\partial \xi} \right) = 0, \quad \text{in } 0 < \xi < l_1, \quad (16)$$

where  $\epsilon_s$  and  $k_s$  denote respectively the solid internal energy and the solid thermal conductivity.

To take into account the thermal diffusion of the grain arising from the grain–grain contact, we complete Eq. (16) with the diffusion term following the  $x$ -direction:

$$\begin{aligned} \frac{\partial \epsilon_s}{\partial t} - \frac{\partial}{\partial x} \left( k_s \frac{\partial \theta_s}{\partial x} \right) - \frac{\partial}{\partial \xi} \left( k_s \frac{\partial \theta_s}{\partial \xi} \right) &= 0, \\ \text{in } 0 < x < L, \quad 0 < \xi < l_1. \end{aligned} \quad (17)$$

The rectangular domain  $\Omega = ]0, L[ \times ]0, l_1[$  represents the domain of the porous media such that the  $x$ -direction is the macroscopic distribution in the fuse and the  $\xi$ -direction is the microscopic distribution in the grain.

We close the model by adding the one-dimensional gas equations of the mass, momentum and energy conservation:

$$\frac{\partial(\rho_g \phi)}{\partial t} + \frac{\partial(\rho_g \phi u)}{\partial x} = 0, \quad (18)$$

$$\frac{\partial(\rho_g \phi u)}{\partial t} + \frac{\partial(\rho_g \phi u^2 + \phi P)}{\partial x} = -\phi^2 \frac{\mu}{K} u - \phi^3 \beta \rho_g |u| u, \quad (19)$$

$$\frac{\partial(E\phi)}{\partial t} + \frac{\partial[\phi u(E + P)]}{\partial x} = -Q + \frac{\partial}{\partial x} \left( k_{\text{geff}} \frac{\partial T_g}{\partial x} \right). \quad (20)$$

To evaluate the heat source term  $Q$  in the gas, we use the same experimental correlation, i.e.

$$Q = h_{\text{sf}} A_0 (T_g(x) - \theta_s(x, l_1)), \quad (21)$$

where  $\theta_s(x, l_1)$  is the temperature at the surface grains.

In the solid, we impose the condition  $\partial \theta_s / \partial x = 0$  at  $\xi = 0$  due to the symmetry at the grain center. At the fluid–solid interface, we set the Neumann condition:

$$k_s \frac{\partial \theta_s}{\partial \xi} = l_2 Q, \quad (22)$$

to describe the heat transfer. To characterize  $l_1$  and  $l_2$ , we integrate Eq. (17) as a function of  $\xi$  and we get with the boundary conditions:

$$\frac{\partial}{\partial t} \int_0^{l_1} \epsilon_s d\xi - \frac{\partial}{\partial x} \left( k_s \frac{\partial}{\partial x} \left( \int_0^{l_1} \theta_s d\xi \right) \right) = l_2 Q. \quad (23)$$

Assuming that we reach the thermally homogeneous case (i.e.  $\epsilon_s$  and  $\theta_s$  do not depend on  $\xi$ ), we have

$\epsilon_s(x, \xi) = e_s(x) = c_{vs} T_s(x)$  and  $\theta_s(x, \xi) = T_s(x)$  and Eq. (23) yields:

$$\frac{\partial e_s}{\partial t} - \frac{\partial}{\partial x} \left( k_s \frac{\partial T_s}{\partial x} \right) = Q \frac{l_2}{l_1}. \quad (24)$$

In comparison with Eq. (8), we have  $l_1 = (1 - \phi)l_2$  where  $l_2$  is the characteristic length of the porous medium and we get  $k_{\text{seff}} = (1 - \phi)k_s$ . In our case, we choose the grain mean radius  $R$  for  $l_2$ .

### 3.1. Numerical method

The numerical method is based on a fractional step technique. Assuming that we know approximations  $\rho_g^n$ ,  $u^n$ ,  $E^n$  and  $T_g^n$  for the gas at time  $t^n$  and  $\theta_s^n$ ,  $\epsilon_s^n$  for the solid and let  $\Delta t$  be a small time interval.

- We first compute an approximation of the flux  $Q^n$  between the gas and the solid phase.
- Secondly, we compute an approximation  $\rho_g^{n+1}$ ,  $u^{n+1}$ ,  $E^{n+1}$  and  $T_g^{n+1}$  solving Eqs. (18)–(20) using the fractional step method described in Section (2.2.2).
- Finally, we solve the heat equation in the solid domain adding at the solid–gas interface the flux  $Q$  to obtain  $\epsilon_s^{n+1}$  and  $\theta_s^{n+1}$  using a finite volume method as presented in the next section.

### 3.2. The heat equation resolution

We consider a two-dimensional mesh composed of rectangular cells. For each cell,  $x_j$ ,  $\xi_j$  denote the cell center coordinates of length  $\Delta x$  and height  $\Delta \xi$ . Since we have to handle two meshes, the one-dimensional mesh for the gas and the two-dimensional mesh for the porous media, we impose a mesh compatibility such that each 2D mesh column corresponds to a 1D cell (see Fig. 4).

The computation is based on an implicit finite volume scheme. Indeed, an explicit scheme requires the use of very small time steps to satisfy the stability condition

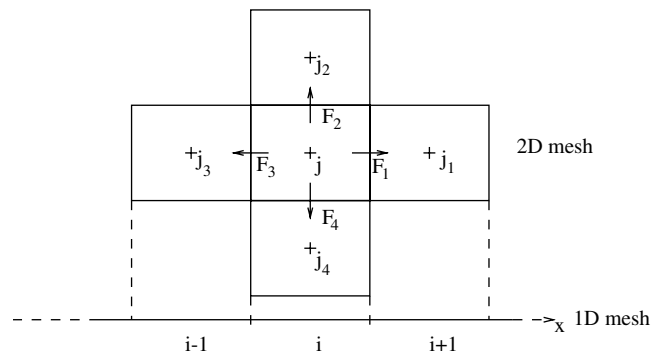


Fig. 4. The control volumes.

because  $\Delta\xi$  is very small which leads to unrealistic computation times.

The implicit formulation is given by:

$$\begin{aligned} \rho_s c_{vs} \theta_{s,j}^{n+1} &= \rho_s c_{vs} \theta_{s,j}^n + \frac{\Delta t}{\Delta x} (F_1^{n+1} - F_3^{n+1}) \\ &+ \frac{\Delta t}{\Delta \xi} (F_2^{n+1} - F_4^{n+1}), \end{aligned} \quad (25)$$

where  $F_1$ ,  $F_2$ ,  $F_3$  and  $F_4$  represent the numerical fluxes at the cell interfaces using a centered difference scheme:

$$\begin{aligned} \rho_s c_{vs} \theta_{s,j}^{n+1} &= \rho_s c_{vs} \theta_{s,j}^n + 2\Delta t k_{j,j_1} (\theta_{s,j_1}^{n+1} - \theta_{s,j}^{n+1}) \frac{(l_{j,j_1})^2}{S_j(S_j + S_{j_1})} \\ &- 2\Delta t k_{j,j_3} (\theta_{s,j_3}^{n+1} - \theta_{s,j}^{n+1}) \frac{(l_{j,j_3})^2}{S_j(S_j + S_{j_3})} \\ &+ 2\Delta t k_{j,j_2} (\theta_{s,j_2}^{n+1} - \theta_{s,j}^{n+1}) \frac{(l_{j,j_2})^2}{S_j(S_j + S_{j_2})} \\ &- 2\Delta t k_{j,j_4} (\theta_{s,j_4}^{n+1} - \theta_{s,j}^{n+1}) \frac{(l_{j,j_4})^2}{S_j(S_j + S_{j_4})}, \end{aligned} \quad (26)$$

where  $l_{jk}$ ,  $k = j_1, j_2, j_3, j_4$  represents the length between the cells and  $S_j$  the  $j$  cell surface.

Due to the boundary condition assumption, we take the flux equal to zero at  $x = 0$ ,  $x = L$ ,  $\xi = 0$ . At  $\xi = l_1$ , we set  $F_2 = l_2 h_{sf} A_0 (T_{g,i}^n - \theta_{s,j}^n)$  where the cell  $j$  is the upper 2D mesh cell of the column  $i$  which correspond to the 1D cell  $i$ .

The discretized heat equation is rewritten in the matrix form:

$$A\theta_s^{n+1} = B(\theta_s^n), \quad (27)$$

where  $A$  is a symmetrical matrix since we have a regular mesh. We obtain  $\theta_s^{n+1}$  solving the linear system Eq. (27) by a conjugated gradient method.

### 3.3. Convergence test

To test the convergence of the numerical scheme, we use an 80 cell mesh following the  $x$ -direction and four

discretizations of 10, 20, 40 and 80 cells following the  $\xi$ -direction to see the mesh dependence on the heat transfer between the gas and the solid (Fig. 5). The initial and boundary conditions are identical to the ones prescribed in the previous section.

Numerical simulations show that we obtain a good stability result for an  $80 \times 80$  cell mesh and the numerical solutions converge to a limite as we refine the mesh.

## 4. Comparison of the two models

We compare the models in two characteristic situations: the first one consists of slowly heating a porous medium for 1 s and the second one consists of quickly heating the porous medium for 5 ms to highlight the difference between the two models. In the following calculations, we use the sand grain characteristics:  $l_1 = 180 \mu\text{m}$ ,  $k_s = 1.8 \text{ W m}^{-1} \text{ K}^{-1}$ ,  $c_p = 1420 \text{ J kg}^{-1} \text{ K}^{-1}$  and  $\rho = 2650 \text{ kg m}^{-3}$ .

### 4.1. Heat exchanger simulation

A constant mass flux  $(\rho_g \phi u)_0 = 25 \text{ kg m}^{-2} \text{ s}^{-1}$  at temperature  $(T_g)_0 = 1180 \text{ K}$  enters for 1 s into the porous medium at  $x = 0$ . We compare the evolution of the velocity, density, gas and solid temperatures for the two models. For the 1-model, we have used an 80-cell mesh and for the 2-model a  $80 \times 80$  cell mesh.

Fig. 6 presents the comparison of the velocity and the density distributions for the gas between the two models at  $t = 1 \text{ s}$ . We observe that the velocity and density profiles calculated with the two models almost coincide. Fig. 7 presents the comparison of the gas temperature and the solid temperature distributions at  $t = 1 \text{ s}$ . For the 2-model, we show the temperature interface at  $\xi = l_1$  and the temperature core at  $\xi = 0$  in a sand grain. We observe that the solid temperature is homogeneous, i.e. the interface and core temperatures are equal. In this case, the gas and the solid phases are in thermal

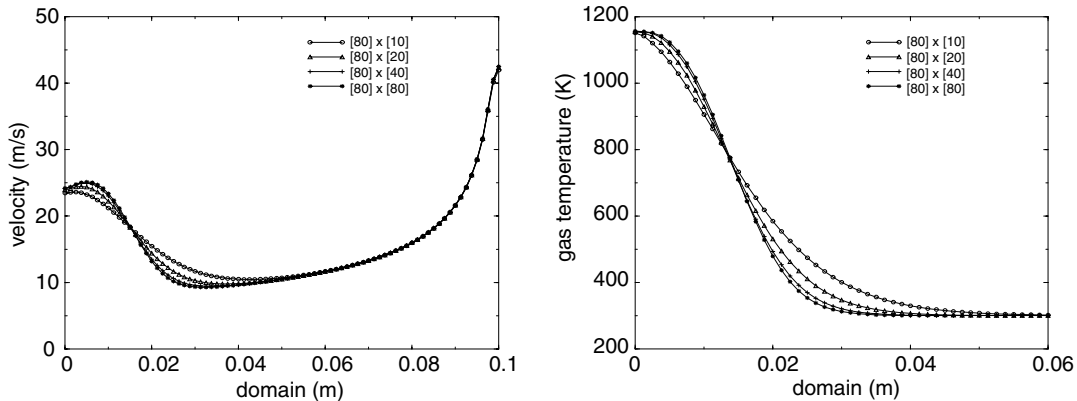


Fig. 5. Velocity and temperature distributions in the porous medium domain at  $t = 1 \text{ s}$  for several meshes.



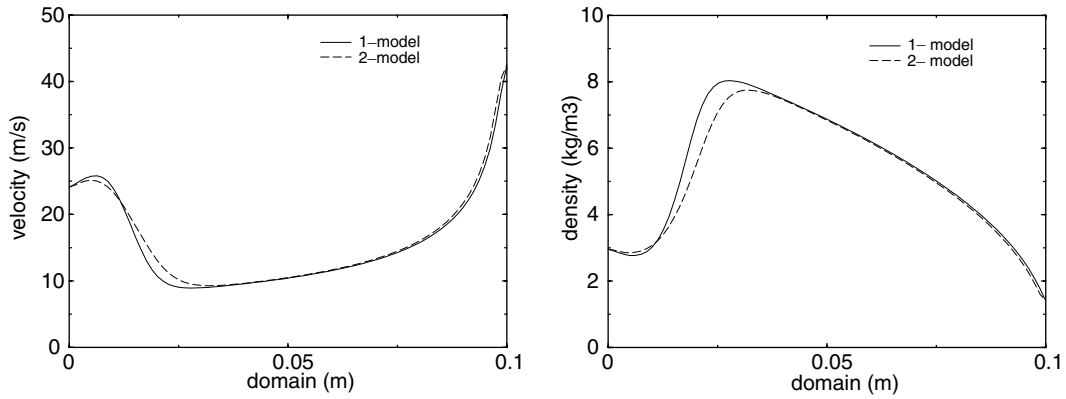


Fig. 6. Comparison of velocity and density distributions in the porous medium domain at  $t = 1$  s.

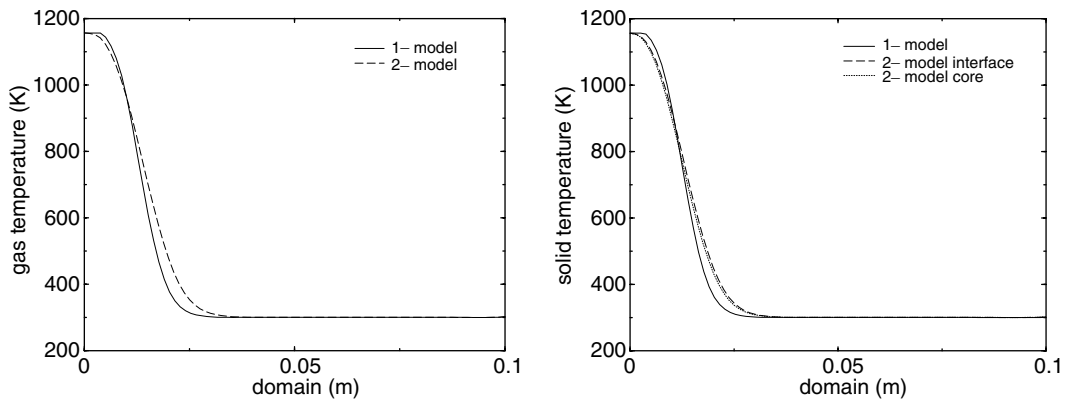


Fig. 7. Comparison of the gas and the solid temperature distributions in the porous medium domain at  $t = 1$  s.

equilibrium  $T_g \approx \theta_s$  for the two models and we note that the curves are similar with a slight difference in the heat transfer zone. The difference derives from the small temperature difference between the core and the interface grain. The surface temperature is a somewhat greater than the temperature in the first model and the heat transfer is less important: we transport the energy for a little longer.

#### 4.2. The HBC fuse simulation (5 ms)

The second test concerns the explosive case. A constant inlet flux of gas is injected into the porous medium with  $(\rho_g \phi u)_0 = 70 \text{ kg m}^{-2} \text{ s}^{-1}$  and a gas temperature of  $(T_g)_0 = 3000 \text{ K}$ . Computations have been performed for a period of 5 ms with a  $L = 10 \text{ mm}$  domain length.

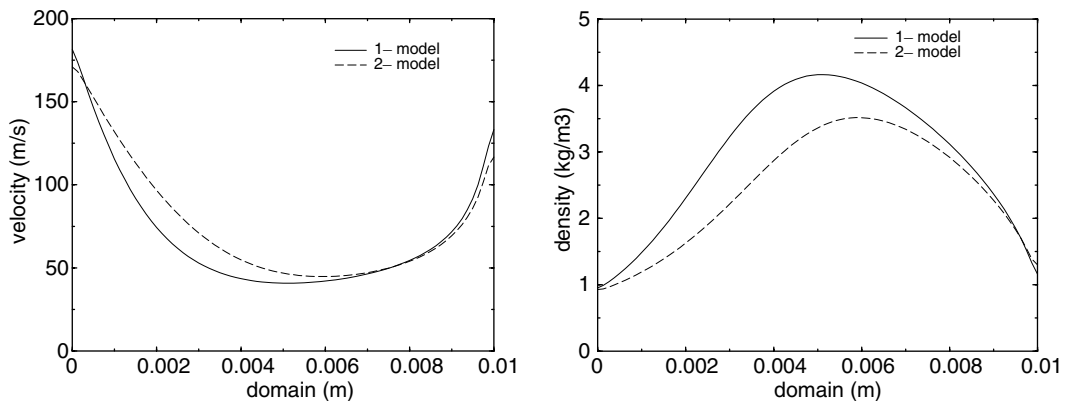


Fig. 8. Comparison of the velocity and density distributions in the porous medium domain at  $t = 5$  ms between the two models.

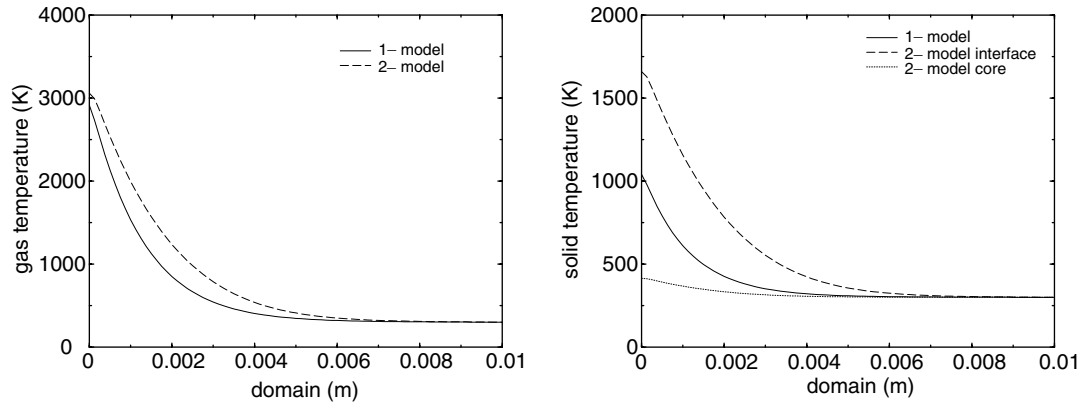


Fig. 9. Comparison of the gas and solid temperature distributions in the porous medium domain at  $t = 5$  ms between the two models.

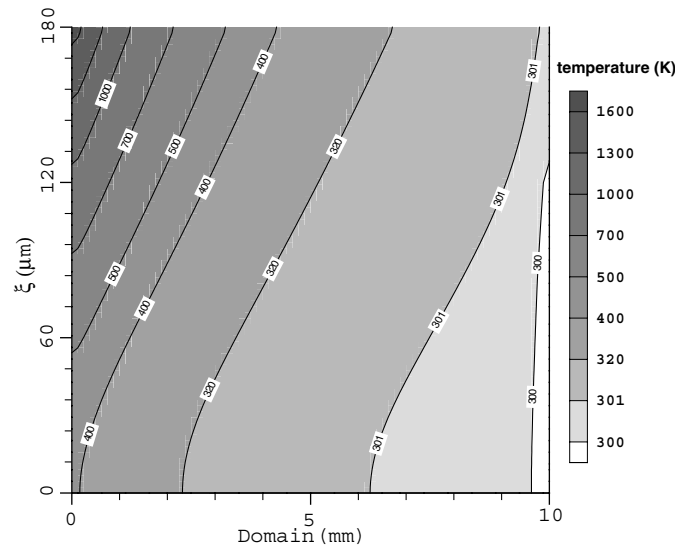


Fig. 10. Temperature distribution in the porous medium following  $x$ -direction and in the sand grains following the  $\xi$ -direction at time  $t = 5$  ms.

Fig. 8 presents the velocity and density of the gas obtained from the two models at  $t = 5$  ms. In this particular situation, the velocity and density profiles of the 2-model are different from the 1-model. The mechanical effect of the gas flow is also modified, in particular the Darcy and Forchheimer force distribution. Fig. 9(left) presents the comparison of the gas temperature and the Fig. 9(right) presents the solid temperature obtained from the 1-model and the interface, core solid temperatures of the 2-model. The heat transfer with the 1-model is greater than that with the 2-model. As explained in the previous section, the energy is transported further since the heat transfer is lower in the 2-model than the 1-model. This numerical experiment shows that the solid homogeneous temperature assumption is no longer valid. The interface and the core temperature inside a grain are dramatically different since we obtain an interface temperature of around 1650 K while the core temperature is 420 K.

Fig. 10 represents the temperature distribution in the porous medium at time  $t = 5$  ms. The  $\xi$ -direction gives the distribution in the grain for any fixed  $x$ . We obtain a large temperature differential at  $x = 0$  which confirms the non-homogeneous distribution of the energy inside the grains and modifies the heat transfer. This phenomenon occurs during the fuse operation process where the first grains are partially vaporized.

## 5. Conclusion

In this paper, we have proposed a new modelling of a compressible gas flow in porous media based on a microscopic description of the heat transfer between the two-phases. We have used a one-dimensional gas flow model in porous media taking into account the mechanical interaction between the gas and the porous medium. To evaluate the heat transfer between the

two phases, we have introduced a two-dimensional local heat transfer model depending on the characteristic length  $l_1$  of the medium.

Numerical simulations using a fractional step technique and a finite volume method have been performed to evaluate the solid temperature distribution in silica sand grains. We have compared the model with a classical thermal non-equilibrium homogenized model for two characteristic situations. We have shown that the two models agree when the gas inflow velocity is low and when the simulation time is long. However, for the explosion situation, the assumption that we have a solid homogeneous temperature does not hold when the diffusion into the solid is less rapid than the whole process duration. The 1-model is computationally less costly and should be used in the case of a homogeneous temperature distribution in the grain. The 2-model only has to be considered when the temperature gradient is important, in particular when the vaporization process occurs. In this case, the thermal distribution will be helpful to determine the vaporization rate.

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